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Introduction

Aim: Identify the relaxation pathways of Glycolaldehyde upon laser irradiation

Glycolaldehyde

- Possible source of OH radicals
- Product of alkenes-ozone reactions in atmosphere
- Isomer of acetic acid and methylformate, found in the interstellar medium

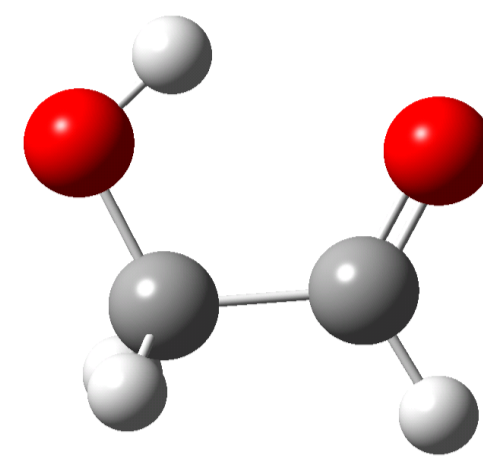
Issues

- Fragmentation vs. isomerisation
- Identify chemical routes to form complex organic species in interstellar molecular clouds [1]

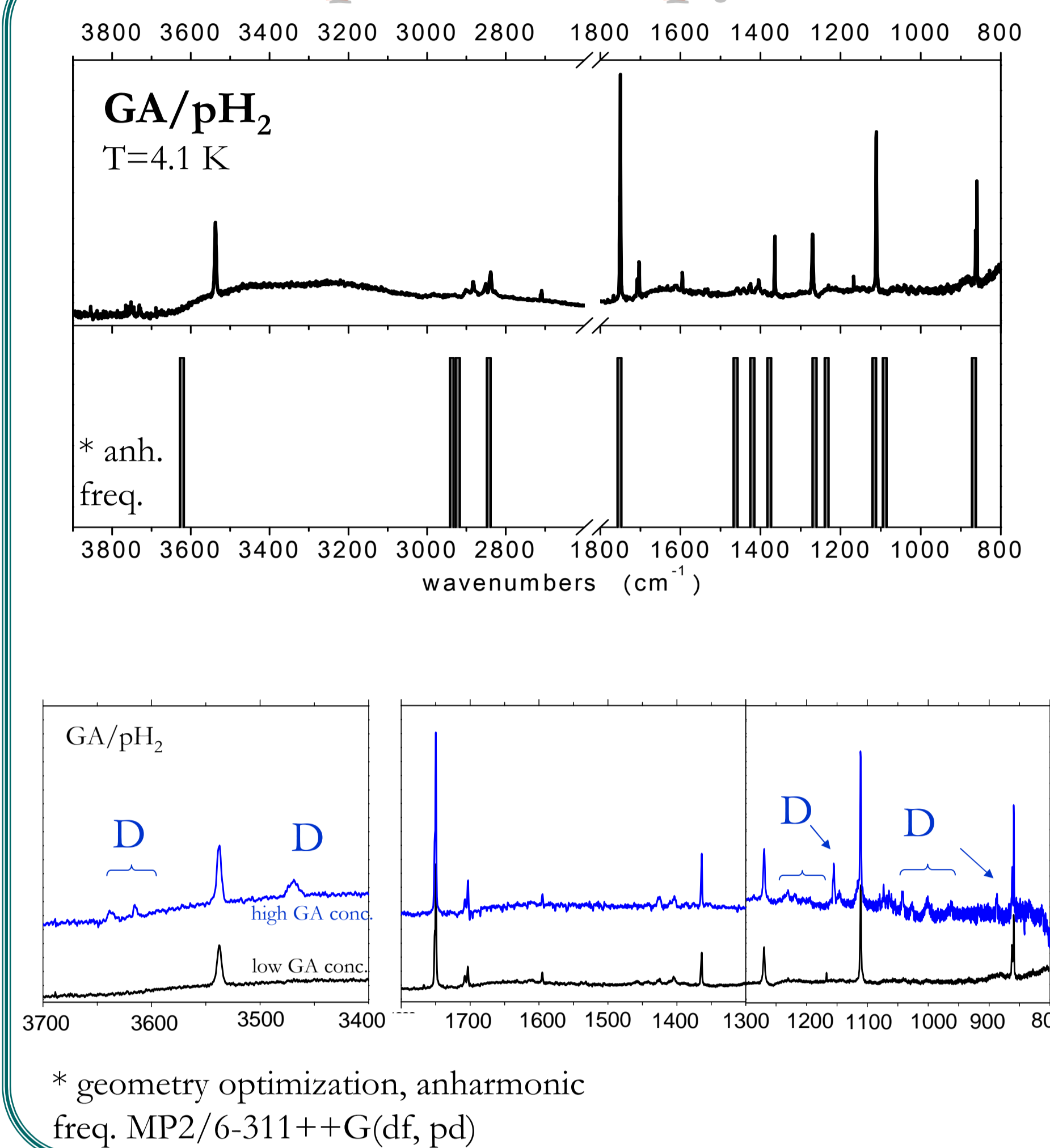
Methods

Matrix isolation spectroscopy
IR and UV irradiation
Photoproducts monitored by IR spectroscopy

cis-cis conformer



IR spectroscopy



GA/pH₂ [2]

- Ar and pH₂ matrices
- Only Cis-cis (Cc) conformer observed
- Reduced site effect in pH₂ ⇒ well isolated bands
- Anharmonic calc. ⇒ Fermi resonances (bands re-assignment)

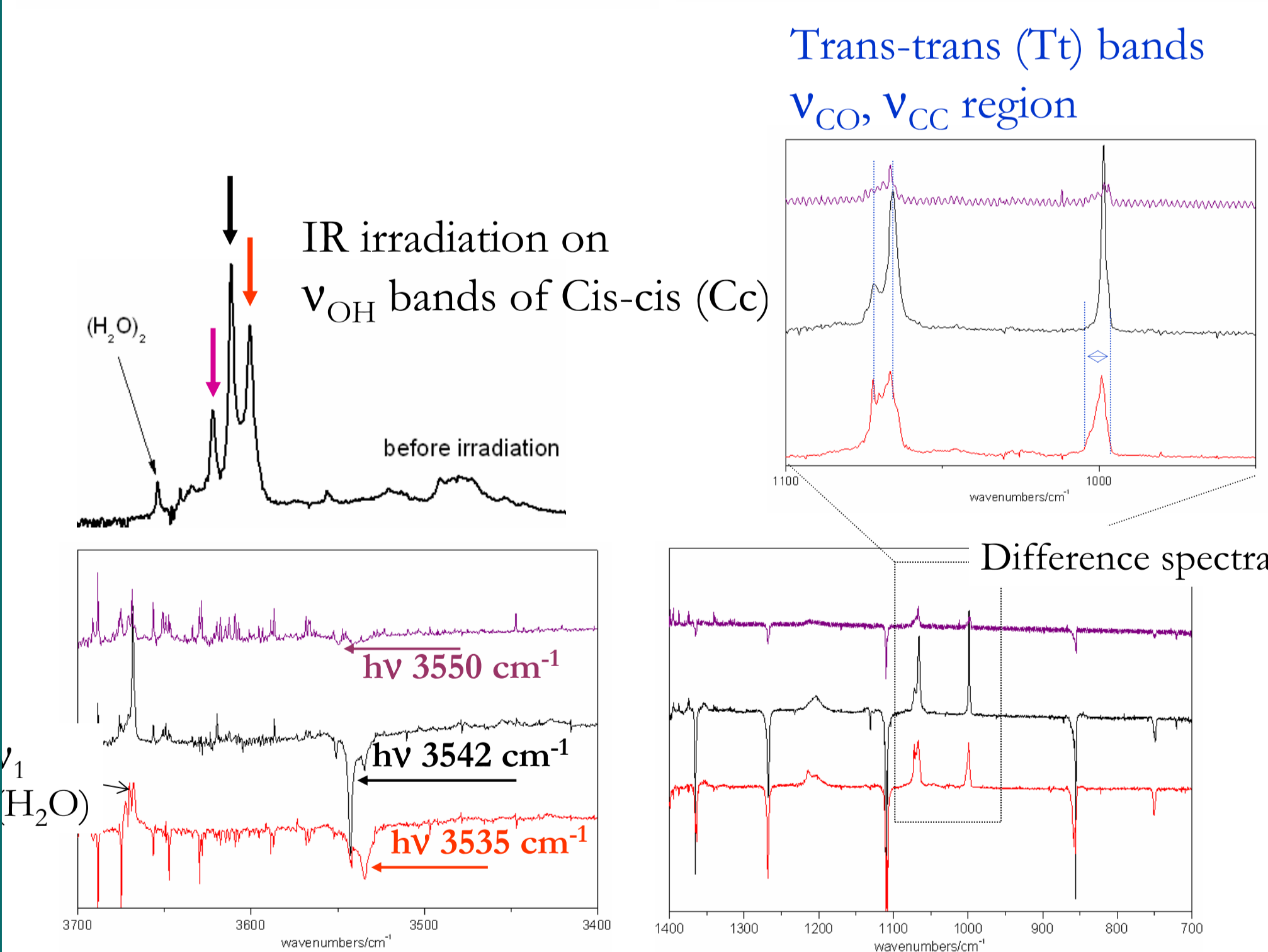
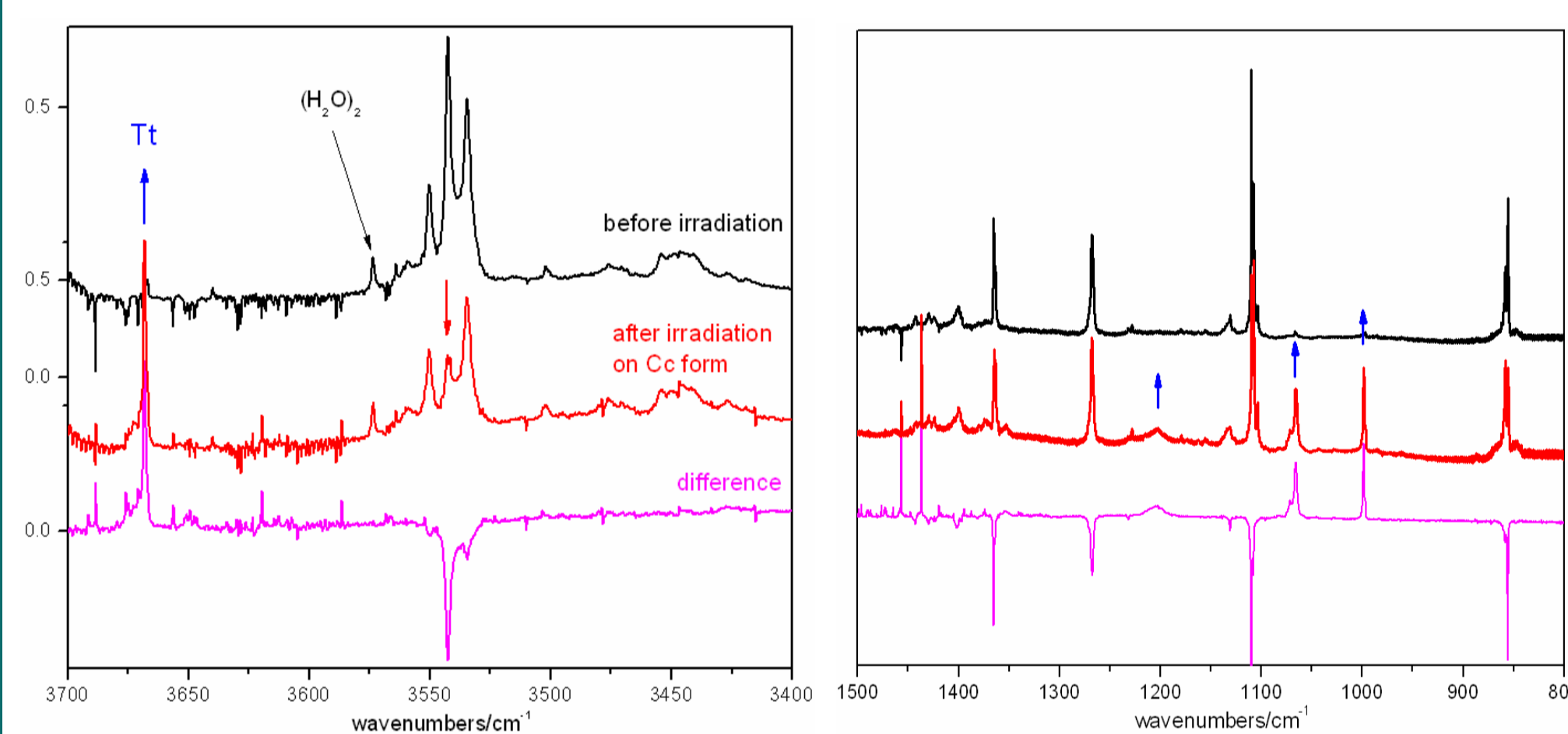
Dimers

- Band identification
- Need calculations

GA-H₂O complex (under way)

IR irradiation GA/Ar

- ps IR OPO laser ($\Delta\nu \sim 2\text{cm}^{-1}$)
- IR selective excitation
- ⇒ OH stretch
- ⇒ different sites

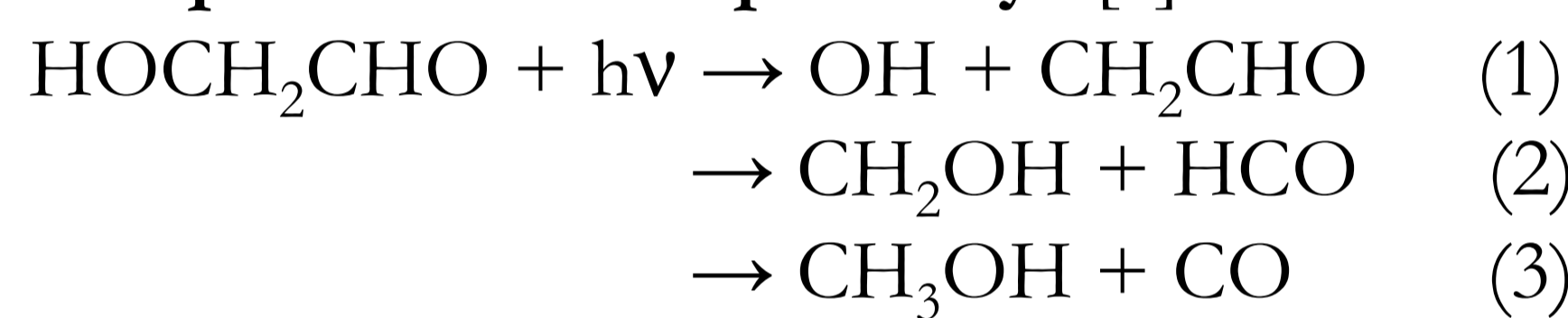


- ✓ Isomerisation Cc ⇒ Tt [3]
- ✓ Tt ⇒ Cc reverse process less efficient
- ✓ Excitation of other vibrational modes

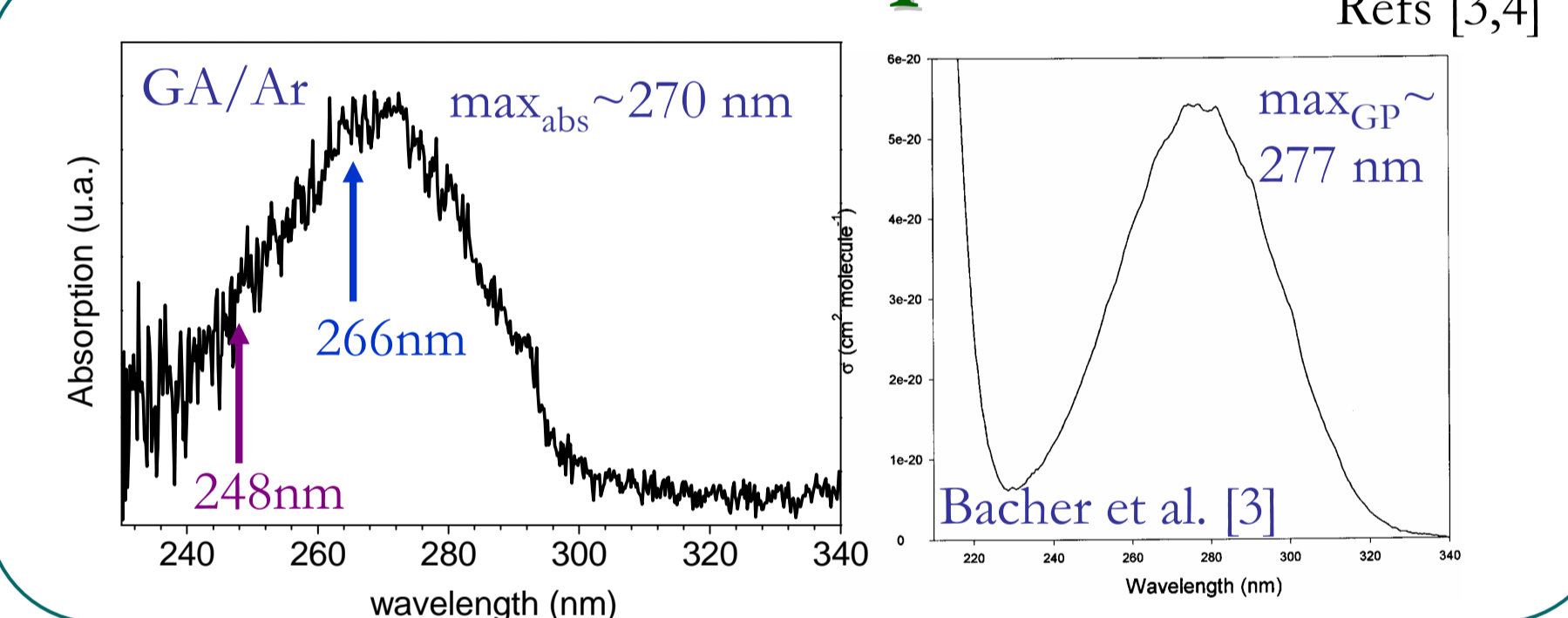
UV irradiation

- Three excitation energies: 266, 248, 193 nm

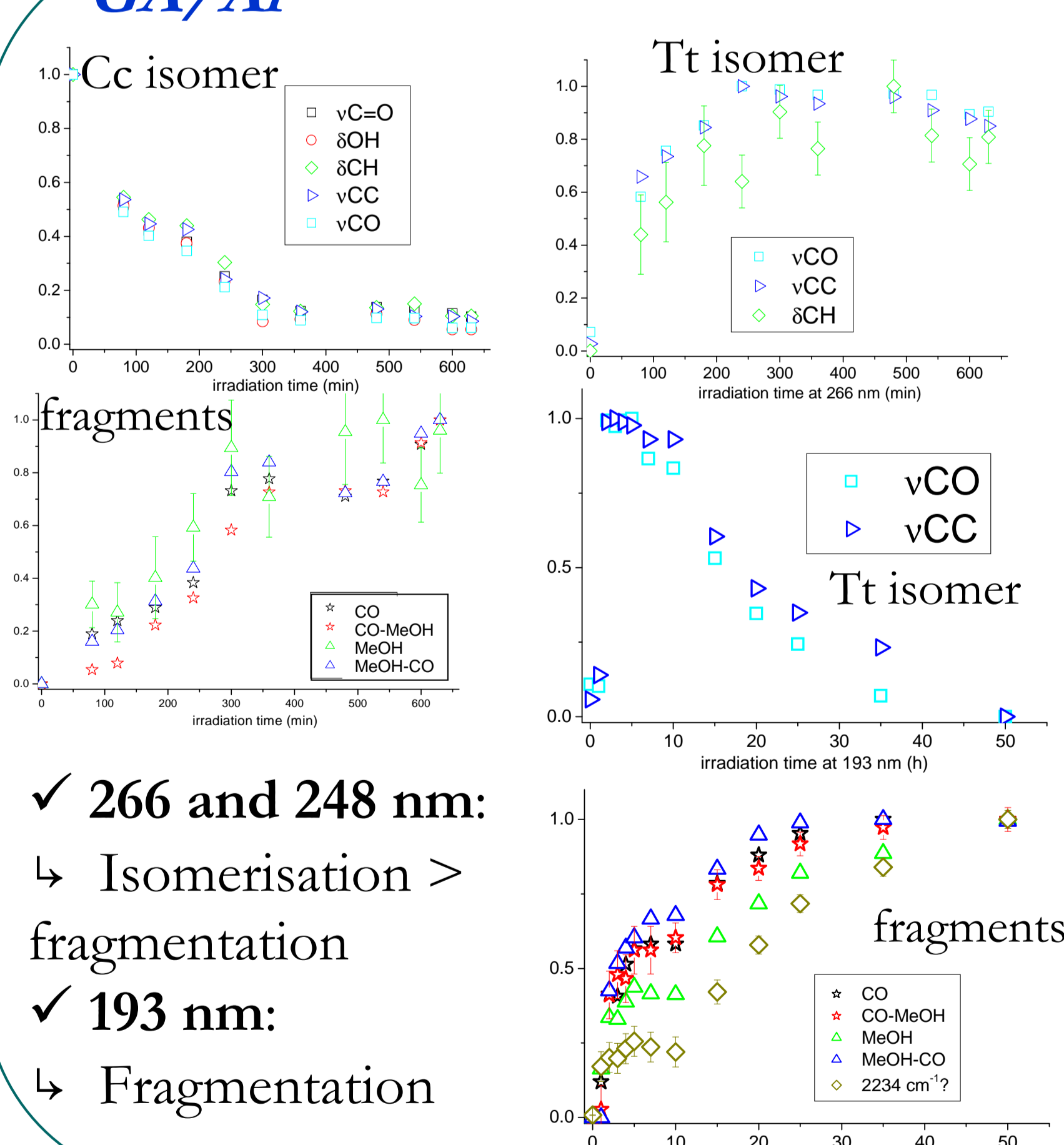
Gas phase reaction pathways [3]:



UV absorption



GA/Ar Kinetics

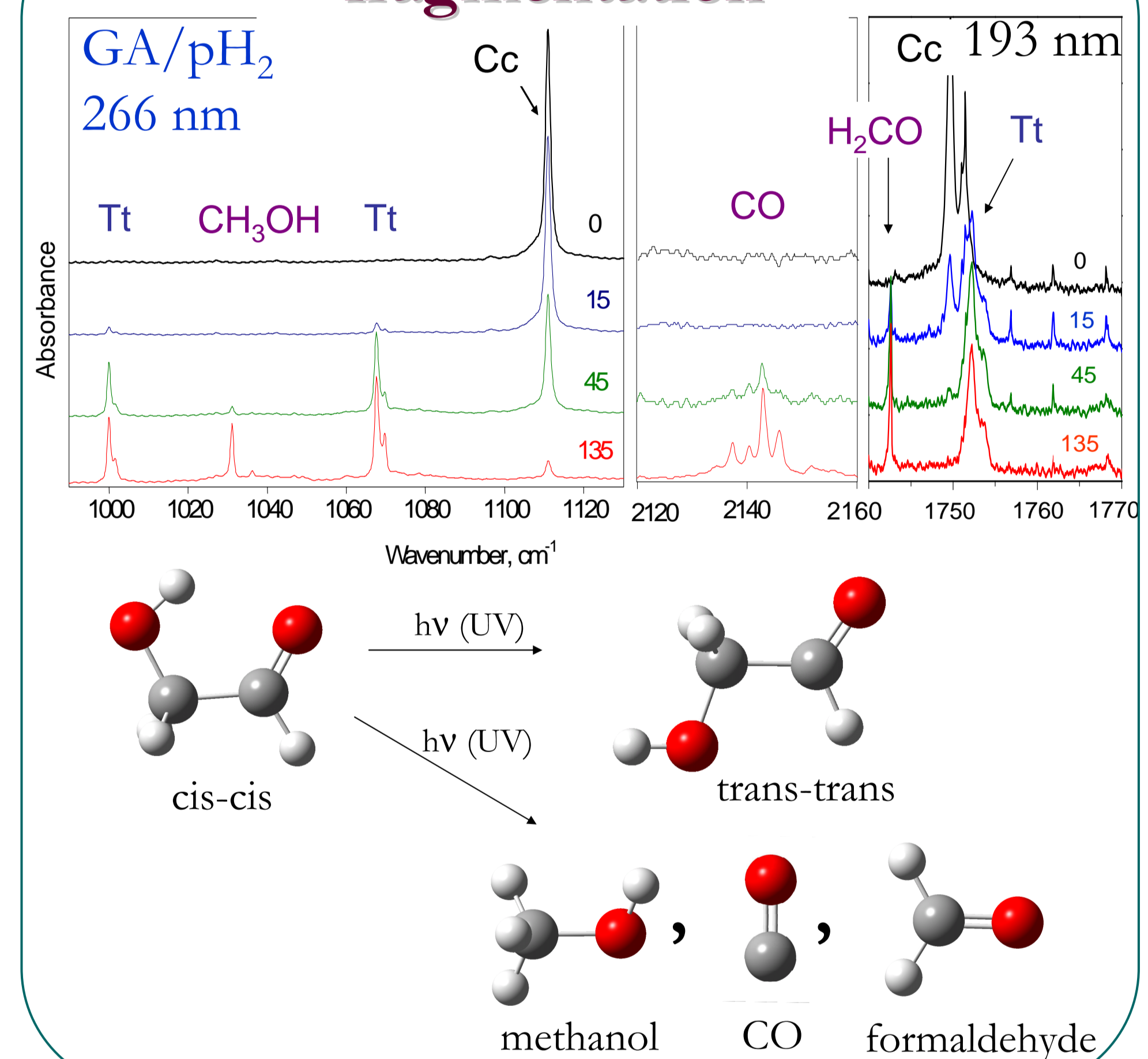


- ✓ 266 and 248 nm:
↳ Isomerisation > fragmentation
- ✓ 193 nm:
↳ Fragmentation

Outlooks

- Excited states calculations
- Vibrational dynamics
- ⇒ Effect of the environment (Photon echo studies)
 - Different matrices (rare gas, pH₂, D₂, etc.)
 - Solvation effect (dimers, H₂O-complexes)
 - Effect of H-bonding

Isomerisation vs. fragmentation



- Isomerisation & fragmentation
- New fragments at 193 nm
- Different dissociation mechanism at 193 nm

Gas phase [3]: H₂CO, CO, CO₂, CH₃OH, HCOOH

Ar and pH₂: CO, CO-CH₃OH (Ar), CH₃OH, CH₃OH-CO (Ar), H₂CO (193 nm) + Tt conformer

IR & UV-induced isomerisation

- Two conformers observed among the four predicted [5]
- Isomerisation Cis-cis ⇒ Trans-trans
- Reverse process Trans-trans ⇒ Cis-cis only induced by IR
- UV irradiation
 - Competition between isomerisation and fragmentation
 - Dependence with hν_{exc} ⇒ excitation of different electronic states
 - No radicals

Acknowledgements

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References :

- [1] C.J. Bennett and R.I. Kaiser, Ap. J. 661, 899 (2007). [2] J.Ceponkus, W. Chin, M. Chevalier, A. Limongi, M. Broquier, and C. Crépin, J.Chem. Phys. (2010) accepted. [3]C. Bacher, G. S. Tyndall, and J. J. Orlando, J. Atm. Chem. 39, 171 (2001), I. Magoner, A. Mellouki, G. Le Bras, G.K. Moortgat, A. Horowitz, J. Phys. Chem. A 109,4552 (2005). [3] A. Aspiala, J. Murto and P. Stén. Chem. Phys. 106, 399 (1986). [4] A. Beeby, D. B. H. Mohammed, and J.R. Sodeau, J. Am. Chem. Soc. 109, 857 (1987). [5] Senent et al., J. Phys. Chem.A 108, 6286 (2004).